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SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c)

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Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide.

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## **Answers**

The answers to the Hold chemistry chapter 12 section 1 review can not be located online Chapter 9 modern chemistry review answers. Students will have to obtain assistance from the teacher if If you are referring to first question of the section review for chapter 6. 1 in the Addison-Wesley

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chemistry book, there is a table on the same page as. . .

## **Chapter 9 Modern Chemistry Review Answers**

CHAPTER 7 REVIEW Chemical Formulas  
and Chemical Compounds SECTION 2  
SHORT ANSWER Answer the following  
questions in the space provided. 1.

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Assign the oxidation number to the specified element in each of the following examples:

4 a. S in  $\text{H}_2\text{SO}_3$  3 6  
b. S in  $\text{MgSO}_4$  2 c. S in  $\text{K}_2\text{S}$  1 d. Cu in  $\text{Cu}_2\text{S}$  6  
e. Cr in  $\text{Na}_2\text{CrO}_4$  5 f. N in  $\text{HNO}_3$  4  
g. C in  $(\text{HCO}_3)_2$  ...

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